organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Yi-Min Hu, Yi-Zhi Li, Jie Zhou, Hong-Wen Hu and Yi Pan*

Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

Correspondence e-mail: llyyjz@nju.edu.cn

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.040 wR factor = 0.115 Data-to-parameter ratio = 13.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Methyl {2-[4-ethyl-1-(tolylsulfonyl)-2,3dihydropyrrol-3-yl]ethyl}benzoate

The title compound, $C_{23}H_{27}NO_4S$, is a novel kind of pyrrole derivative. The molecules are connected by weak $C-H\cdots O$ interactions, forming one-dimensional chains along the *b* axis.

Received 4 September 2003 Accepted 23 September 2003 Online 30 September 2003

Comment

Heterocyclic nuclei are present in many pharmacologically active compounds (Tsuji, 1995). Many synthetic methods for the construction of cyclic compounds have been developed (Schuster & Blechert, 1997). Palladium-catalysed annulation technology has become one of the most powerful tools in the construction of heterocyclic compounds (Beletskaya & Cheprakov, 2000). Aryl or olefinic halides have been used as starting materials in various cases, but a few examples of palladium-catalysed annulation reactions of alkyl halides are also known. In our continuing research on palladium-catalysed Heck reactions of benzyl halides with olefins (Hu, Zhou, Long *et al.*, 2003; Hu, Zhou, Lian *et al.*, 2003), a novel palladium-catalysed reaction of benzyl halides with a diene is reported here.



The molecular structure of the title compound, (I), is shown in Fig. 1. In the ethylene link between two rings, the C3–C7 and C7–C8 bond lengths are 1.580 (3) and 1.442 (3) Å, respectively. These values deviate markedly from the typical Csp^3-Csp^3 bond distance (1.54 Å). This may be an artifact due to dynamic disorder, created by pedal motion as observed in stilbenes and azobenzenes (Harada & Ogawa, 2001).

There is a chiral carbon atom, C3, in the molecule, but the title compound is racemic. The five-membered ring, comprising atoms C1–C4/N1, is nearly planar, the r.m.s. deviation being 0.025 Å. Atoms C1, C4, S1 and N1 are not coplanar, suggesting sp^3 hybridization of atom N1. There are weak intermolecular C–H···O interactions (Table 2 and Fig. 2), which lead to one-dimensional molecular chains along the *b* axis.

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved



Figure 1

A view of the molecule of (I), showing 50% probability displacement ellipsoids.

Experimental

The title compound, (I), was synthesized according to the following procedure: an oven-dried Schlenk flask was evacuated and filled with nitrogen, then charged with N-allyl-N-(2-butenyl)-p-toluenesulfonamide (1.325 g, 5 mmol), 3-methoxycarbonylbenzyl chloride (1.562 g, 5.5 mmol), tributylamine (1.5 ml, 6 mmol), Pd(OAc)₂ (12 mg, 0.05 mmol) and DMF (2 ml), giving a yellow solution. The reaction mixture was heated at 393 K with stirring. The reaction mixture was cooled to room temperature after 15 h and the resultant yelloworange mixture was diluted with Et₂O (10 ml). The mixture was washed with H₂O (15 ml) and the aqueous layer was extracted with Et_2O (10 ml). The combined organic layers were dried (MgSO₄), filtered, and concentrated in vacuo. The crude material was purified by flash column chromatography on silica gel (petroleum ether/ EtOAc, 6:1) and recrystallized from MeCN/Et₂O; yield 0.97 g (47%). Colourless crystals of (I) suitable for X-ray diffraction were obtained by diffusion of ethyl acetate into an MeCN solution over a period of one week.

Crystal data

C II NO C	7 2
$C_{23}H_{27}NO_4S$	$\mathbf{Z} = \mathbf{Z}$
$M_r = 413.52$	$D_x = 1$
Triclinic, P1	Mo K
a = 8.412 (2) Å	Cell p
b = 10.931 (2) Å	refl
c = 11.989 (2) Å	$\theta = 2.2$
$\alpha = 83.73 \ (1)^{\circ}$	$\mu = 0.$
$\beta = 81.30 \ (1)^{\circ}$	T = 29
$\gamma = 80.32 \ (1)^{\circ}$	Prism
$V = 1070.2 (4) \text{ Å}^3$	$0.3 \times$

Data collection

Bruker SMART APEX CCD area-	R
detector diffractometer	θ_1
φ and ω scans	h
5318 measured reflections	k
3680 independent reflections	l
3009 reflections with $L > 2\sigma(I)$	

 $D_x = 1.283 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation Cell parameters from 620 reflections $\theta = 2.7 - 20.8^{\circ}$ $\mu = 0.18~\mathrm{mm}^{-1}$ T = 293 (2) KPrism. colourless $0.3 \times 0.2 \times 0.2$ mm

CCD area-	$R_{\rm int} = 0.017$
er	$\theta_{\rm max} = 25.0^{\circ}$
	$h = -9 \rightarrow 9$
ons	$k = -12 \rightarrow 12$
ctions	$l = -14 \rightarrow 10$
a (r)	



Figure 2

The crystal packing of (I), viewed down the a axis, showing, as dashed lines, the C-H···O short contact. [Symmetry codes: (A) 2 - x, 1 - y, 1-z; (B) 2-x, -y, 1-z; (C) 2-x, -1-y, 1-z.]

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.065P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	+ 0.0934P]
$wR(F^2) = 0.115$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.06	$(\Delta/\sigma)_{\rm max} < 0.001$
3680 reflections	$\Delta \rho_{\rm max} = 0.41 \text{ e } \text{\AA}^{-3}$
265 parameters	$\Delta \rho_{\rm min} = -0.41 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

\$1-O1	1.4218 (15)	C1-C2	1.322 (3)
S1-O2	1.4249 (14)	C2-C3	1.503 (3)
\$1-N1	1.6321 (18)	C3-C4	1.533 (3)
S1-C17	1.757 (2)	C3-C7	1.580 (3)
N1-C1	1.418 (3)	C7-C8	1.442 (3)
N1-C4	1.492 (2)		
C1-N1-C4	107.12 (14)	C8-C7-C3	114.56 (18)
C1-N1-S1	118.77 (13)	C7-C8-C9	108.61 (18)
C4-N1-S1	118.38 (14)	C10-C9-C14	118.25 (19)
C2-C3-C4	104.18 (15)	C10-C9-C8	116.2 (2)
C2-C3-C7	106.42 (17)	C14-C9-C8	125.50 (19)
C4-C3-C7	114.10 (18)		

Table 2		
Hydrogen-bonding geometry	(Å,	°).

	ם ת	Ш Л	D 4		
D=II···A	$D=\Pi$	11····A	$D \cdots A$	$D=\prod \cdots A$	
$C4-H4A\cdots O1^{i}$	0.97	2.54	3.488 (2)	166	
C1-H1···O2 ⁱⁱ	0.93	2.65	3.523 (2)	157	

Symmetry codes: (i) 2 - x, -y, 1 - z; (ii) 2 - x, 1 - y, 1 - z.

All H atoms were positioned geometrically and refined as riding, with $U_{\rm iso}({\rm H})$ values 1.2 to 1.5 times $U_{\rm eq}$ of the parent atom. The distances to the H atoms were in the range 0.93–0.98 Å.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

We thank the National Science Foundation of China (project No. 20072015) and the National 863 High Technology Program for financial support. This work was also supported by the Nanjing University Talent Development Foundation (No. 0205005122).

References

- Beletskaya, I. P. & Cheprakov, A. V. (2000). Chem. Rev. 100, 3009-3066.
- Bruker (2000). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Harada, J. & Ogawa, K. (2001). J. Am. Chem. Soc. 123, 10884–10888.
- Hu, Y.-M., Zhou, J., Lian, H.-Z., Zhu, C.-J. & Pan, Y. (2003). Synthesis, pp. 1177–1180.
 Hu, Y.-M., Zhou, J., Long, X.-T., Han, J.-L., Zhu, C.-J. & Pan, Y. (2003).
- Tetrahedron Lett. **44**, 5009–5010.

Schuster, M. & Blechert, S. (1997). Angew. Chem. Int. Ed. Engl. 36, 2037–2056. Tsuji, J. (1995). Palladium Reagents and Catalysis: Innovations in Organic Synthesis. pp. 182–226. Chichester: Wiley.